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Crowne Plaza Hotel Baton Rouge, LA

### High-Temperature Stable, Hydrophobic Coatings for Antifouling Applications

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### **Outline and Rationale**

- 1. Emissions Reduction
  - Global need to lower emissions reduced fouling rates may reduce CO<sub>2</sub>
- 2. Benefits of Hydrophobic Coatings
  - Lowers surface energy, % surface polarity (SP). May protect against corrosion, oxidation & carburization
- 3. Laboratory Experiments
  - Fouling rate reduction, measurements of surface energy & polarity
- 4. Industrial Field Trials
  - Vacuum wash beds, Heat exchangers



Rare Earth Coating – Minimox<sup>®</sup> Alloy Treatment

- Not a traditional coating, more like surface doping.
- Water-based product yields scattering of rare earth nanoparticles. Not continuous coating.
- After oxidation, a superalloy is created on the surface of more basic alloys.

• Minimox<sup>®</sup> is a trademark of Material Interface, Inc.

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# Hydrophobic Coatings in Process Equipment

Rare Earth Coating – Technical Benefits

- After thermal processing, yields a low surface energy coating.
- Creates a thin, robust surface modified layer.
- High temperature stable. Temperature limits imposed by alloy, not the coating process.
- Inside and outside surfaces can be treated dipping or spraying application.
- Suspended nanoparticles. No VOC. Non-hazardous. pH 7. Surfactant is present.

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# Hydrophobic Coatings in Process Equipment

Rare Earth Coating – Historical Development

- Process created to improve high temperature oxidation.
- High temperature alloys contain rare earths for stable oxide layers.
- Water-based nanoparticles applied to surface have the same effect as rare earths added to the alloy, at significant cost savings.
- For oxidation resistance, good efficacy on stainless steel, nickel, aluminum, magnesium, and titanium alloys.



### Rare Earth Coating – Oxidation Resistance Examples

Simple processing – Amazing results



Thermocouple protection tube 304 stainless 200 hrs; 1500°F (815°C)





### Rare Earth Coating – Oxidation Resistance Examples **Turbine Blade Heat Treating – 403 SS, Air, 1850°F (1010°C)**

Asreceived



After Nano rare earth coating

Heat treated in air without rare earths



Heat treated in air with RE pretreatment



### Rare Earth Coating – Durable

Coat flat stock; air dry	Bend	Successful protection



Rare Earth Coating – Durable





Rare Earth Coating – Durable





### Rare Earth Coating – Different surface compounds







Rare Earth Coating – Different surface compounds – How?





Rare Earth Coating – Different surface compounds – How?





Oxidation without Rare Earths

Oxidation with Rare Earths

Metal diffuses through surface layer to react with oxygen Oxygen diffuses through surface layer to form oxide

Presence of rare earths –

- Changes direction of oxide layer formation
- Keeps the rare earths on the surface of the alloy
- Effectively halts diffusion in/out



Initial tests at Chevron (538°C, 1000°F) indicated rare earth coatings:

- Reduced fouling by 80% in laboratory test rigs
- Reduced adhesion of coke to the test sample





**Rare Earth Coatings** 

Initial tests at Chevron indicated rare earth coatings reduced fouling in laboratory test rigs.

Why?

Surface energy measurements indicated the treated areas were hydrophobic. Is this the reason for reduced fouling? What other parameters may be relevant?





Rare Earth Coatings – Surface energy measurements



HIGH CONTACT ANGLE Low surface energy, Good release, non-stick properties HYDROPHOBIC



LOW CONTACT ANGLE High surface energy, Poor release, Poor non-stick properties HYDROPHILIC



Rare Earth Coatings – Multiyear Research Project Matrix of alloys tested: Nickel Alloy 200 (UNS N02200), Hastelloy C276 (UNS N10276) 304, 316, 317, 347, and 410 Stainless Steels, Inconel 800 (UNS N08800), Inconel 825 (UNS N08825), Duplex 2205 (UNS S32305) F5 (UNS K41545), F9 (UNS K90941), Titanium Grade 2 (UNS R50400), 1008 carbon steel, aluminum bronze (UNS C95400).

Surface energy measured as a function of post-coating processing time and temperature. Also SEM/EDS and XPS.

System used for most of the surface energy measurements was a Krüss MSA (Mobile Surface Analyzer). Experiments verified with a Ramé-Hart Model 590 automated tensiometer at Chevron/Richmond.



#### Rare Earth Coatings – Surface energy measurements





Automatically doses with water and diiodomethane droplets



#### Rare Earth Coatings – Krüss MSA & Software

The MSA doses two parallel drops, followed by the direct analysis of the contact angles and the derived results of the surface free energy. The specific mathematical correction program is that of Owens-Wendt-Rabel-Kaelble (OWRK).



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## Hydrophobic Coatings in Process Equipment

### Rare Earth Coatings – Krüss MSA & Software

SFE calculation			
Model OWRK -	Substance	Mean CA [º]	
Use error weighting	o 🔽 water	105.46 (±12.33)	
	x 🔽 diiodo-methane	57.98 (±9.29)	
Correlation coefficient : 1.00 $\frac{(1+\cos\theta)\cdot\sigma_1}{2\sqrt{\sigma_1^D}}$ 16 12	Calculation was so Surface free energy Disperse	uccessful. 29.77 ±5.73 mN/m 29.74 ±5.34 mN/m	
8 - 4 - <b>C</b>	Polar	0.05 20.56 mitym	
0			
0 0.4 0.8 $1.2 \frac{1.6}{\sqrt{\sigma_1^P/\sigma_1^D}}$			

The MSA software yields the total Surface Free Energy (SFE) and also the Disperse and Polar portions.

Total SFE = Disperse + Polar fractions

Disperse – Weaker van der Waals bonds.

Polar – Tight chemical bonds to adherents.

% Surface Polarity = Polar fraction/total SFE



Rare Earth Coatings – Multiyear Research Project Surface energy measured f(post-coating time/temperature) for wide range of alloys.

Total SFE and % surface polarity measured.

Test coupons usually had a mill oxide surface and a ground metallic surface on the reverse side.

What is in the coating? Generally  $Y_2O_3$  and a surfactant.



#### Finding: Rare earth coatings reduce surface energy



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### Hydrophobic Coatings in Process Equipment

#### Finding: Rare earth coatings reduce surface polarity to near 0%





### Rare Earth Coatings – Multiyear Research Project

316 stainless (mill oxide finish), Rare earth treated



Water contact angle = 113° Surface energy = 26.4 mN/m; % Surface polarity = 0.0%

*316 stainless (mill oxide finish), Untreated* 



Water contact angle = 54° Surface energy = 49.6 mN/m; % Surface polarity = 22.1%



Rare Earth Coatings – Multiyear Research Project

**Key Conclusions** 

- Essentially all alloys showed a minimum surface energy after processing to 400°C for one hour.
- F5 and F9 more difficult, but ultimately successful.
- Carbon steel also successful.
- After heating to 400°C for one hour, the surface polarity approaches 0%. The treated surfaces become nonpolar.



Rare Earth Coatings – Multiyear Research Project

- What is special about 400°C?
  - The surfactant burns off and a thin thermal oxide is grown
- What is special about Y<sub>2</sub>O<sub>3</sub>?
  - Surface energy is intrinsically related to a material's work function. The work function of  $Y_2O_3$  is very low.
  - Catalytic activity?
- How thick is the thermal oxide +  $Y_2O_3$ ?
  - About 100 nm



#### Rare Earth Coatings – Multiyear Research Project

What do the samples look like? Thin gold thermal oxide is present. QC conducted with x-ray fluorescence (XRF) for yttrium.



(More obvious differences in previous slides at higher temperatures/long time.)



Treated

Untreated

# Hydrophobic Coatings in Process Equipment

Rare Earth Coatings – Ramifications for Process Equipment

Because the layer is thin and integral to the alloy, it cannot spall/flake off.

Because it is thin, there is negligible additional thermal resistance associated with the treatment.

The surface layer cannot flake off with thermal cycling or bending.

The process is high-temperature stable and does not impose any temperature limitations beyond those of the base metal.

Surfaces generated are nonpolar.





Rare Earth Coatings – Ramifications for Process Equipment

### Coating complex structures

- Rare earth coating can be applied to tortuous & complicated geometries
- Can be applied to low-clearance surfaces and structures that are not generally amenable to coating with other traditional systems.
- Treated surface can later be bent and formed with no major restrictions

If it can be wetted with water, it can be treated.



### Rare Earth Coatings – Processing Treated 317 stainless vacuum column wash bed



Two isopropanol wash tanks, blow off residues, dip in rare earth suspension, allow to drain, air dry, heat to 400°C for one hour.



### Rare Earth Coatings – Vacuum Column Wash Bed



After dipping/air drying After treatment + 400° High surface area

In service for three years without showing any pressure drop buildup due to coking.



Rare Earth Coatings – Heat Exchangers



After treatment



Finned tubes

• Installed Summer 2022; expectations of improvement in ease of cleaning, thermal efficiency and reduction in fouling rates. Treated inside and outside tubing surfaces.



### Rare Earth Coatings – Heat Exchangers

Unexpected event caused rapid fouling. Removed after  $\approx$  6 mos.





Exchanger cleaned successfully with hydroblasting. Rare earths still present; reinstalled in service without recoating.  $Y_2O_3$  verified still present with x-ray fluorescence.



### Rare Earth Coatings – Heat Exchangers

Another unexpected event in same system again caused rapid fouling.

Tube sheet looked poor, but residue very loosely adhering.



Exchanger easily cleaned successfully with 12,500 psig hydroblast. Rare earths still present; reinstalled in service without recoating.



#### Rare Earth Coatings – Heat Exchangers

Analysis of large foulant flake from tube sheet Machining lines from tube sheet visible on foulant indicating clean separation.

FTIR indicated the surface against the tube sheet and the exposed surface contained identical organic species. No yttria detected with XPS.

SURFACE FREE ENERGY (SFE) AND % SURFACE POLARITY OF SYSTEM:

Foulant in contact with tube sheet = 41 mN/m SFE, 0.4% polarity

Tube sheet = 24 mN/m SFE, 0.0% polarity

Two nonpolar surfaces in contact yield no adhesion.





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# Hydrophobic Coatings in Process Equipment

### Value of Rare Earth Coatings Downstream Applications

- Reduced costs
  - Costs of fouling and decoking significantly lowered
  - Longer times between cleanings; easier cleaning
  - Still present after hydroblasting; no need to reapply
- Risk reduction
  - Fewer shutdowns
  - Lowers probability of safety incidents occurring

- Lowers production losses
  - Fewer unplanned shutdowns
- Better thermal efficiencies
  - Avoids thermal inefficiency
  - Reduced furnace firing
  - Less cooling needs
- Ability to process more challenging feedstocks
- Fine structures can be maintained (finned tubes, textured wash bed inserts)

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# Hydrophobic Coatings in Process Equipment

### Off-shore, Subsea Processing Plants

Rare-earth coated stainless steel coupons were exposed to a salt-waterbased subsea farm to determine the level of corrosion and bioadhesion.

- Rare-earth treated components showed less bioadhesion/corrosion than uncoated and Teflon-coated samples.
- Rare-earth coated aluminum bronze alloys, used in marine applications because of good mechanical properties and corrosion resistance, showed improved hydrophobicity and 0% surface polarity after treatment.
- Research on-going.



Deployment

Production quantities available

International facilities for coating/heating/deployment





#### **Research and Development Team**

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